

# SAFETY DATA SHEET

Revision Date 16-Mar-2019

Version 1

# 1. IDENTIFICATION OF THE SUBSTANCE/PREPARATION AND OF THE COMPANY/UNDERTAKING

Product identifier **Product Name** 

Magnesium Chloride Anhydrous Cakes (from Zirconium Production)

Other means of identification Product Code Synonyms

SAC049 Kroll reduction salt cakes from Zirconium production (Product #106)

Recommended use of the chemical and restrictions on use **Recommended Use** Chemical intermediate. Uses advised against

Details of the supplier of the safety data sheet **Manufacturer Address** ATI, 1000 Six PPG Place, Pittsburgh, PA 15222 USA Emergency telephone number **Emergency Telephone** Chemtrec: 1-800-424-9300

## 2. HAZARDS IDENTIFICATION

#### Classification

This chemical is not considered hazardous by the 2012 OSHA Hazard Communication Standard (29 CFR 1910.1200)

#### Label elements

**Emergency Overview** 

Appearance Large cakes

Physical state Solid

Odor Odorless

## **Precautionary Statements - Prevention**

Do not breathe resulting gases Wear protective gloves/protective clothing/eye protection Protect from moisture

**Precautionary Statements - Response** In case of fire: Use salt (NaCI) or class D dry powder for extinction

## **Precautionary Statements - Storage**

Store in a dry place

# **Precautionary Statements - Disposal**

Dispose of contents/container to an approved waste disposal plant

#### Hazards not otherwise classified (HNOC)

## Not applicable

## Other Information

If this magnesium chloride salt is dissolved in water, handle the insoluble residue as a pyrophoric material, which is especially hazardous when moist, i.e. 5-20% water. Moist residue can ignite with explosive force. Do not accumulate any quantity of this residue.

When mixed with water, heat, steam, and possibly hydrogen and hydrogen sulfide gas may be generated. Do not mix magnesium chloride with water except in a well-ventilated area, under conditions where heat and any gas that may evolve can easily dissipate.

## **3. COMPOSITION/INFORMATION ON INGREDIENTS**

#### Synonyms

Kroll reduction salt cakes from Zirconium production, (Product #106).

Chemical Name	CAS No.	Weight-%
Magnesium Chloride	7786-30-3	94- >99
Magnesium Oxide	1309-48-4	0-5
Water	7732-18-5	0-1
Magnesium	7439-95-4	0-0.3
Zirconium	7440-67-7	0.1

## **4. FIRST AID MEASURES**

## First aid measures In case of contact with eyes, rinse immediately. If eye irritation persists, get medical Eye contact advice/attention. **Skin Contact** Wash off immediately with soap and plenty of water. If fumes from reactions are inhaled, move to fresh air immediately. Inhalation IF SWALLOWED. Call a POISON CENTER or doctor/physician if you feel unwell. Ingestion Most important symptoms and effects, both acute and delayed Symptoms None anticipated. Indication of any immediate medical attention and special treatment needed Treat symptomatically. Note to physicians 5. FIRE-FIGHTING MEASURES

## Suitable extinguishing media

Non-combustible.

Unsuitable extinguishing media If a fire occurs in the area, avoid water contact with the product to prevent evolution of hazardous gases.

<u>Specific hazards arising from the chemical</u> Non-combustible.

Hazardous combustion productsNot applicable.

Explosion data Sensitivity to Mechanical Impact None. Sensitivity to Static Discharge None.

#### Protective equipment and precautions for firefighters

Firefighters should wear self-contained breathing apparatus and full firefighting turnout gear.

6. ACCIDENTAL RELEASE MEASURES

Personal precautions, protective equipment and emergency procedures		
Personal precautions	Use personal protective equipment as required.	
For emergency responders	Use personal protective equipment as required.	
Environmental precautions		
Environmental precautions	Not applicable to massive product.	
Methods and material for containment and cleaning up		
Methods for containment	Not applicable to massive product.	
Methods for cleaning up	Sweep or shovel material into dry containers. Avoid creating uncontrolled dust. Wash the spill location thoroughly with water - remaining magnesium chloride residue would cause the floor to become slippery.	

## 7. HANDLING AND STORAGE

#### Precautions for safe handling

Advice on safe handling If this magnesium chloride salt is dissolved in water, handle the insoluble residue as a pyrophoric material, which is especially hazardous when moist, i.e. 5-20% water. Moist residue can ignite with explosive force. Do not accumulate any quantity of this residue. When mixed with water, heat, steam, and possibly hydrogen and hydrogen sulfide gas may be generated. Do not mix magnesium chloride with water except in a well-ventilated area, under conditions where heat and any gas that may evolve can easily dissipate.

#### Conditions for safe storage, including any incompatibilities

Water.

Storage ConditionsMagnesium chloride solutions in uncoated steel tanks may activate the metal surface so<br/>that when the tanks are drained the surfaces rust quickly consuming available oxygen. Use<br/>safe tank entry procedures with good ventilation and oxygen level monitoring.

Incompatible materials

## 8. EXPOSURE CONTROLS/PERSONAL PROTECTION

#### Control parameters

#### Exposure Guidelines

Chemical Name	ACGIH TLV	OSHA PEL	
Magnesium Chloride 7786-30-3	-	-	
Magnesium Oxide 1309-48-4	TWA: 10 mg/m <sup>3</sup> inhalable fraction	TWA: 15 mg/m <sup>3</sup> fume, total particulate	
Water 7732-18-5	-	-	
Magnesium 7439-95-4	-	-	
Zirconium 7440-67-7	STEL: 10 mg/m <sup>3</sup> STEL: 10 mg/m <sup>3</sup> Zr TWA: 5 mg/m <sup>3</sup> TWA: 5 mg/m <sup>3</sup> Zr	TWA: 5 mg/m <sup>3</sup> Zr (vacated) STEL: 10 mg/m <sup>3</sup> (vacated) STEL: 10 mg/m <sup>3</sup> Zr	

#### Appropriate engineering controls

**Engineering Controls** 

Avoid generation of uncontrolled particles.

Individual protection measures, such as personal protective equipment

Eye/face protection	When airborne particles may be present, appropriate eye protection is recommended. For example, tight-fitting goggles, foam-lined safety glasses or other protective equipment that shield the eyes from particles.
Skin and body protection	Wear protective gloves.
Respiratory protection	When particulates/fumes/gases are generated and if exposure limits are exceeded or irritation is experienced, proper approved respiratory protection should be worn. Positive-pressure supplied air respirators may be required for high airborne contaminant concentrations. Respiratory protection must be provided in accordance with current local regulations.
General Hygiene Considerations	Handle in accordance with good industrial hygiene and safety practice.

# 9. PHYSICAL AND CHEMICAL PROPERTIES

## Information on basic physical and chemical properties

Physical state Appearance Color	Solid Large cakes Grey silver	Odor Odor threshold	Odorless Not applicable
<u>Property</u> pH Melting point/freezing point	<u>Values</u> - 710 °C / 1310 °F	Remarks • Method Not applicable	
Boiling point / boiling range Flash point Evaporation rate Flammability (solid, gas)	- - -	Not applicable Not applicable Not flammable	
Flammability Limit in Air Upper flammability limit: Lower flammability limit:	-	Not hannable	
Vapor pressure Vapor density Specific Gravity	- - 2.3	Not applicable Not applicable	
Water solubility Solubility in other solvents Partition coefficient Autoignition temperature	-	Not applicable Not applicable	
Decomposition temperature Kinematic viscosity Dynamic viscosity	-	Not applicable Not applicable	
Explosive properties Oxidizing properties Other Information	Not applicable Not applicable		
Softening point Molecular weight VOC Content (%) Density	- - Not applicable -		
Bulk density	100-110 lb/ft3		

# **10. STABILITY AND REACTIVITY**

Reactivity Reacts with water

## **Chemical stability**

Stable under normal conditions.

## **Possibility of Hazardous Reactions**

Reacts with water.

Hazardous polymerization Hazardous polymerization does not occur.

#### Conditions to avoid

Unintentional contact with water. When mixed with water, heat, steam, and possibly hydrogen and hydrogen sulfide gas may be generated. Do not mix magnesium chloride with water except in a well-ventilated area, under conditions where heat and any gas that may evolve can easily dissipate.

#### Incompatible materials Water.

## Hazardous Decomposition Products

None while dry and cool. Magnesium chloride heated above 110°C in the presence of moisture will evolve hydrogen chloride fumes.

## **11. TOXICOLOGICAL INFORMATION**

#### Information on likely routes of exposure

#### **Product Information**

Inhalation	Not an expected route of exposure for product in massive form.
Eye contact	Not an expected route of exposure for product in massive form.
Skin Contact	Product not classified.
Ingestion	Not an expected route of exposure for product in massive form.

Chemical Name	Oral LD50	Dermal LD50	Inhalation LC50
Magnesium Chloride	5000 mg/kg bw	>2000 mg/kg bw	-
7786-30-3			
Magnesium Oxide	-	-	-
1309-48-4			
Water	-	-	-
7732-18-5			
Magnesium	>2000 mg/kg bw	-	-
7439-95-4			
Zirconium	> 5000 mg/kg bw	-	>4.3 mg/L
7440-67-7			-

#### Information on toxicological effects

Symptoms

None known.

Delayed and immediate effects as well as chronic effects from short and long-term exposure

Acute toxicity	Product not classified.
Skin corrosion/irritation	Product not classified.
Serious eye damage/eye irritation	Product not classified.
Sensitization	Product not classified.
Germ cell mutagenicity	Product not classified.
Carcinogenicity	Product not classified.
Reproductive toxicity	Product not classified.
STOT - single exposure	Product not classified.
STOT - repeated exposure	Product not classified.

#### Aspiration hazard

Product not classified.

# 12. ECOLOGICAL INFORMATION

## **Ecotoxicity**

This product as shipped is not classified for aquatic toxicity.

Chemical Name	Algae/aquatic plants	Fish	Toxicity to microorganisms	Crustacea
Magnesium Chloride 7786-30-3	The 72 h EC50 of magnesium chloride to Desmodesmus subspicatus was greater than 100 mg of MgCl2/L.	The 96 h LC50 of magnesium chloride to Pimephales promelas was 2119.3 mg of MgCl2/L.	The 3 h EC50 of magnesium chloride for activated sludge was greater than 900 mg of MgCl2/L.	The 48 h LC50 of magnesium chloride hexahydrate to Daphnia magna was 548.4 mg of MgCl2/L.
Magnesium Oxide 1309-48-4	-	-	-	-
Water 7732-18-5	-	-	-	-
Magnesium 7439-95-4	The 72 h EC50 of magnesium chloride hexahydrate to Desmodesmus subspicatus was greater than 12 mg of Mg/L.	The 96 h LC50 of magnesium chloride to Pimephales promelas was 541 mg of Mg/L.	The 3 h EC50 of magnesium chloride hexahydrate for activated sludge was greater than 108 mg of Mg/L.	The 48 h LC50 of magnesium chloride to Ceriodaphnia dubia was 225 mg of Mg/L. The 48 h LC50 of magnesium chloride hexahydrate to Daphnia magna was 322 mg of Mg/L.
Zirconium 7440-67-7	The 14 d NOEC of zirconium dichloride oxide to Chlorella vulgaris was greater than 102.5 mg of Zr/L.	The 96 h LL50 of zirconium to Danio rerio was greater than 74.03 mg/L.	-	The 48 h EC50 of zirconium dioxide to Daphnia magna was greater than 74.03 mg of Zr/L.

## Persistence and degradability

#### Other adverse effects

## **13. DISPOSAL CONSIDERATIONS**

#### Waste treatment methods

Disposal of wastes	Disposal should be in accordance with applicable regional, national and local laws and regulations. Note that USEPA may consider waste magnesium chloride anhydrous to exhibit the hazardous characteristic of reactivity (D003). If this magnesium chloride salt is dissolved in water, handle the insoluble residue as a pyrophoric material, which is especially hazardous when moist, i.e. 5-20% water. Moist residue can ignite with explosive force. Do not accumulate any quantity of this residue. Dispose of residue, which may exhibit the hazardous characteristic of ignitability (D001) and/or reactivity (D003), per Federal, State, and Local requirements.
Contaminated packaging	Disposal should be in accordance with applicable regional, national and local laws and regulations.

This product contains one or more substances that are listed with the State of California as a hazardous waste.

# **14. TRANSPORT INFORMATION**

DOT

Not regulated

## **15. REGULATORY INFORMATION**

International Inventories	
TSCA	Complies
DSL/NDSL	Complies
EINECS/ELINCS	Complies
ENCS	Complies
IECSC	Complies
KECL	Complies
PICCS	Complies
AICS	Complies

Legend:

TSCA - United States Toxic Substances Control Act Section 8(b) Inventory DSL/NDSL - Canadian Domestic Substances List/Non-Domestic Substances List EINECS/ELINCS - European Inventory of Existing Chemical Substances/European List of Notified Chemical Substances

**ENCS** - Japan Existing and New Chemical Substances

**IECSC** - China Inventory of Existing Chemical Substances

**KECL** - Korean Existing and Evaluated Chemical Substances

PICCS - Philippines Inventory of Chemicals and Chemical Substances

AICS - Australian Inventory of Chemical Substances

## US Federal Regulations

#### SARA 313

Section 313 of Title III of the Superfund Amendments and Reauthorization Act of 1986 (SARA). This product does not contain any chemicals which are subject to the reporting requirements of the Act and Title 40 of the Code of Federal Regulations, Part 372

SARA 311/312 Hazard Categories
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Acute health hazard	No
Chronic Health Hazard	No
Fire hazard	No
Sudden release of pressure hazard	No
Reactive Hazard	No

#### **CWA (Clean Water Act)**

This product does not contain any substances regulated as pollutants pursuant to the Clean Water Act (40 CFR 122.21 and 40 CFR 122.42)

#### **CERCLA**

This material, as supplied, does not contain any substances regulated as hazardous substances under the Comprehensive Environmental Response Compensation and Liability Act (CERCLA) (40 CFR 302) or the Superfund Amendments and Reauthorization Act (SARA) (40 CFR 355). There may be specific reporting requirements at the local, regional, or state level pertaining to releases of this material

#### US State Regulations

#### **California Proposition 65**

This product does not contain any Proposition 65 chemicals

#### U.S. State Right-to-Know Regulations

Chemical Name	New Jersey	Massachusetts	Pennsylvania
Magnesium Oxide 1309-48-4	Х	X	Х
Water 7732-18-5			Х

Magnesium 7439-95-4	Х	Х	Х
Zirconium 7440-67-7	Х	Х	Х

## U.S. EPA Label Information

EPA Pesticide Registration Number Not applicable

16. OTHER INFORMATION				
NFPA_	Health hazards 0	Flammability 0	Instability 1	Physical and Chemical Properties -
HMIS Chronic Hazard Star Lege	Health hazards 1* end *= Chronic	Flammability 0 Health Hazard	Physical hazards 1	Personal protection X
Issue Date	15-Mar-20	)19		
Revision Date	16-Mar-20	)19		
Revision Note				
New Safety Data Sheet				
Note:				
The information provid	ed in this safety data s	heet is correct to the be	est of our knowledge, infor	mation and belief at the

date of its publication. The information given is designed only as a guidance for safe handling, use, processing, storage, transportation, disposal and release and is not to be considered a warranty or quality specification. The information relates only to the specific material designated and may not be valid for such material used in combination with any other materials or in any process, unless specified in the text.

#### End of Safety Data Sheet

Additional information available Safety data sheets and labels available at ATImetals.com from: